# GSC Open File D3772

# Reanalysis of Stream Sediments from the Little Nahanni River Map Sheet (105I), Yukon and Northwest Territories

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## Introduction

A reconnaissance geochemical survey of stream sediments and waters in the Little Nahanni River (NTS 105I) map area was carried out in 1981 as part of the Nahanni Integrated Multidisciplinary Pilot Project (NIMPP). The NIMPP was established by the Geological Survey of Canada to focus the efforts of personnel from different geological disciplines (Goodfellow, 1982). In addition to reconnaissance coverage of stream sediment and water geochemistry and bedrock and surficial geology, mineral deposit studies and detailed studies of plutonic rocks were carried out (Gordey & Andersen, 1993). Information derived from these studies was used primarily to determine areas of high potential for Pb, Zn, Ba and W.

Granitoid plutons and their extensive hornfels zones have recently become the focus of gold exploration as a result of two recent exploration successes, Fort Knox in Alaska and Brewery Creek in Yukon (Hart, 1999). Mid-Cretaceous granitoid intrusions, generally referred to as the Tombstone Plutonic Suite (TPS) (Poulsen et al., 1997), form a narrow belt of lithologically diverse plutons extending from Alaska eastwards across Yukon to the Northwest Territories (Lang et al., 1997). Mid-Cretaceous granite and granodioritic intrusions of the Selwyn Plutonic Suite (Gordey & Anderson, 1993) within the map area and in adjacent regions are now included with the TPS, based on location, age and similarities in metallogeny (Duncan et al., 1998). Gordey & Andersen (1993) estimate that 7% of the map area (1051) is underlain by granitoid plutons. Nearly all TPS intrusions contain Au-mineralization (Baker et al., 1999). Mineralization is widely developed in the TPS, with a defining metal assemblage of Au-W-Bi, variable Sn-As-Sb-Mo and minor Pb-Zn-Ag (Lang et al., 1997).

Under the terms of the Canada/Yukon Geoscience Program, the Government of Yukon, the Department of Indian and Northern Affairs and the Geological Survey of Canada agreed to conduct a joint research project consisting of reanalysis of previously collected stream sediments for gold and 25 other elements. The addition of gold and many of the lithophile elements associated with felsic plutons in continental arc environments (Lang et al., 1997) to the existing suite of elements may help identify potential areas of large tonnage-low grade gold mineralization within the Little Nahanni River map area. Data files in this open file include original field and analytical data for stream sediments and waters and new data for stream sediments.

# Physiography

The Little Nahanni River (National Topographic System (NTS) 105I) map area (Fig. 1, Fig. 2) covers 11,489 km<sup>2</sup> (Sebert & Munro, 1972) between latitude 62°N and 63°N and longitude 128°W and 130°W. The border between Yukon Territory and Northwest Territories (District of Mackenzie) divides the map sheet roughly diagonally from northwest to southeast.



**Figure 1** Map of Yukon Territory showing survey area (in yellow) and current GSC open file numbers for geochemical data releases. Coverage consists of stream sediment and water surveys except for GSC Open File 2860 (NTS 105A, Watson Lake) which is a lake sediment and water survey.

The valley of the South Nahanni River separates the Mackenzie Mountains in the northeast from the Selwyn Mountains (and Logan Mountains within the Selwyn Mountains) and Yukon Plateau in the southwest half of the survey area (Fig. 2).



Figure 2Map of survey area (NTS 105I) showing major rivers and geographic features.Areas of textured blue in the southeast indicate glacier cover.

The land surface is rugged and consists mainly of slope, summit and plateau (Jackson, 1987). Elevations range from 762 m to 2590 m (Gordey & Anderson, 1993), and more than 75% of the survey area lies between elevations of 1220

and 1820 m (Jackson, 1987). Glaciers are found at higher elevations of the Ragged Range, in the southeast corner of the map sheet. The timberline is located between elevations 1375 m to 1525 m, below which bedrock exposure is rare (Gordey & Anderson, 1993).

A major drainage divide between the Little Nahanni River and the Pelly, Ross and Hyland Rivers is delineated by the Yukon-Northwest Territories border. Streams in Yukon drain west to southwest and in the Northwest Territories flow southeast towards the Mackenzie River. Thousands of small streams flow from higher elevations, forming dendritic drainage patterns.

# Geology (from Gordey & Anderson, 1993)

The survey area falls within two major depositional environments, the Mackenzie Platform and the Selwyn Basin (Gordey & Anderson, 1993), a late Precambrian to Middle Devonian platform-basin assemblage (Fig. 3). To the southwest, turbiditic sandstone, shale, deep-water limestone and chert of the Selwyn Basin are gradually interstratified moving to the northeast with shallow-water sandstone, dolostone and limestone of the Mackenzie Platform. Euxenic black shale of Early Silurian age is host to important stratiform Pb-Zn deposits.



**Figure 3** Major tectonic and depositional areas in the area of the survey (modified from Duncan et al., 1998).

Late Devonian shale overlies Mackenzie Platform to the northeast. To the southwest, coarse clastics in a number of submarine fan complexes deposited turbiditic quartz-chert sandstone and chert pebble conglomerate derived from elevated fault blocks of older Selwyn Basin strata. Stratiform barite and Ba-Pb-

Zn deposits associated with local faulting form important deposits within black siliceous shale of Middle to Late Devonian age.

Devonian-Mississippian turbiditic clastics are succeeded by mid-Mississippian quartz sandstone and shale and shale, chert, minor sandstone and siltstone of Early Permian and Triassic age.

Granite and granodiorite intrusions of mid-Cretaceous age underlie about 7% of the survey area. Circular in plan, plutons vary between one and 20 km in diameter and intrude and hornfels strata as young as Triassic. Plutons range in age from 88 to 114 Ma. Plutons can be classified into two groups depending on whether or not hornblende is present, or alternatively, whether or not two micas (muscovite and biotite) are present. Tungsten in skarns is associated with two-mica plutons intruding argillaceous limestone.

Metamorphic grade is equivalent to subgreenschist facies.

## Surficial Geology (from Jackson, 1987)

The survey area was intensely glaciated during the Wisconsinan-age McConnell advance. At climax, one or more ice centres existed over the area. Ice flow was directed by the underlying topography.

As noted earlier, little bedrock is exposed below approximately 1500 m. The most common glacial deposits are tills deposited as blankets or veneers over bedrock. Glaciofluvial deposits consisting primarily of gravel form kames, eskers and planar deposits in most valleys. Glaciolacustrine silts and clays are found where glacial ice impeded drainage during glaciation and occur extensively in the South Nahanni River valley, especially near Mt. Wilson. Bogs have formed in areas of poor drainage.

### **Mineral Deposits**

Until recently, mineral exploration was directed towards four types of mineralisation recognized within the survey area:

- 1. Early Silurian or Devonian shale or chert hosted stratiform Pb-Zn;
- 2. replacement Zn-Pb deposits in platform carbonates of the Mackenzie Platform;
- 3. Zn-Pb veins peripheral or within mid-Cretaceous granitic plutons (Bonham-Carter & Goodfellow, 1986);
- 4. W skarn deposits developed within limestone host rocks adjacent to mid-Cretaceous granitic plutons (Gordey & Anderson, 1993).

The Canada Tungsten mine, located immediately south of the southeast corner of the survey area, was once the world's largest producer of scheelite concentrate (Dick & Hodgson, 1982).

Plutons and surrounding host rocks of the Tombstone Plutonic Suite that were originally the target of tungsten exploration are now the focus of exploration for gold. The development of effective heap leach mining methods in cold terrains

and exploration successes at Fort Knox (Alaska) and Brewery Creek (Yukon) have focused recent exploration in Yukon in areas magmatic rocks of the TPS occur (Hart, 1999).

Mineralization is varied and appears to be controlled by the location relative to the intrusion and the nature of the host rock. Within intrusions, sheeted Au quartz vein mineralization occurs, characterized by metal assemblages of Au-Bi-W-As-Te (Mo-Sb) (e.g. Fort Knox, Dublin Gulch) (Baker et al., 1999). Mineralization in hornfels zones includes W skarns and Sn and Au-bearing breccias (e.g. Ray Gulch, MacTung) (Lang et al., 1997). Outside of contact zones Pb-Zn-Ag-Au veins are found (e.g. Keno Hill) as well as Au-bearing disseminated to replacement deposits (e.g. Brewery Creek) (Baker et al., 1999; Lang et al., 1997).

## **Collection Procedures and Sample Management (Original Surveys)**

Stream sediment and water samples at 984 sites were collected at an average density of one sample per 12 km<sup>2</sup> throughout the 11489 km<sup>2</sup> comprising the Nahanni mapsheet (NTS 105I). For the purposes of sampling, preparation and analytical control, sample numbers were divided into blocks of 20, each block consisting of 17 routine samples, one field duplicate, one blind duplicate and one control reference standard. Further details can be obtained from Garrett (1974).

At GSC laboratories in Ottawa, field-dried stream sediment samples were airdried, sieved to minus-80 mesh (177  $\mu$ m) and ball-milled to minus-150 mesh. Control reference and blind duplicate sample positions were filled during sample preparation. In the case of stream waters, the control reference positions were filled in the field with one of three control standards collected near base camp. The blind duplicates for stream waters were left blank.

All stream waters were filtered through 0.45 µm filter paper before analysis.

# Original Analytical Procedures (from Goodfellow, 1982)

### Stream Waters

Uranium was determined by laser-induced fluorescence using a Scintrex UA-3. A 5 ml sample was pipetted into a quartz cell and a metaphosphate-phosphate solution (500  $\mu$ l) added. The fluorescence of the uranyl phosphate formed and excited by the laser was measured. The method of standard additions was used.

Fluoride was measured using a specific ion electrode and an orion meter. A 5 ml sample of water and a 5 ml aliquot of buffer (TISAB) were measured using UV readings. Alkalinity and pH were determined simultaneously using a Radiometer TTT 81 digital titrator and pH meter respectively.

Anions CI, NO<sub>3</sub>, PO<sub>4</sub>, SO<sub>4</sub> were measured using ion chromatography. A 100  $\mu$ I sample was separated on an exchange resin and the resulting solution passed through a high-capacity cation exchange resin to a conductivity cell. The eluent used was 0.003 M Na<sub>2</sub>CO<sub>3</sub>/0.0024 M NaHCO<sub>3</sub>. A Dionex System 12 was used.

Major cations Na, K, Ca, Mg, Mn, Fe, and Zn were determined by direct aspiration using a Perkin-Elmer 5000 atomic absorption spectrophotometer. An air-acetylene flame was used in all cases. A 2000 ppm potassium solution was used as an ionization buffer for Na analyses; 2000 ppm sodium for K analyses, and 2000 ppm lanthanum was used as a releasing agent for Ca analyses.

Table 1 provides a summary of lower detection limits and methods for water analyses.

	ELEMENT	DETECT	TION LEVEL	METHOD
WATERS:				
T-Alk Ca Cl F Fe K Mg Mn Na NO <sup>3-</sup> pH PO <sup>4-</sup> SO <sup>4-</sup> U	Alkalinity Calcium Chloride Fluoride Iron Potassium Magnesium Manganese Sodium Nitrate Hydrogen ion activity Phosphate Sulphate Uranium	2.0 0.5 0.1 25 40 0.2 0.2 10 0.2 0.2 - 0.2 - 0.15 0.5 0.10	ppm ppm ppb ppb ppm ppm ppm - ppm - ppm ppm ppm ppm	TIT AAS IC ISE AAS AAS AAS AAS IC GCM IC IC LIF
Zn	Zinc	5	ppb	AAS

AAS - atomic absorption spectrometry

GCM - glass Calomel electrode and pH meter

IC - ion chromatography

ISE - ion selective electrode

LIF - laser-induced fluorescence

TIT - titration

**Table 1**Summary of analytical data and methods for Waters

### Stream Sediments

Zn, Cu, Pb, Ni, Co, Ag, Cd, Mn, Fe, Mo and V were determined by atomic absorption spectrophotometry after decomposition with a multi-acid total digestion. A 500 mg sample was leached over a period of four hours using a 5-ml aliquot of HF-HCI-HNO<sub>3</sub>-HCIO<sub>4</sub> (ratio 1:1:1:1). A final solution of 5% HCl was made up to a volume of 20 ml. All elements, with the exception of Mo, were atomized using an air-acetylene flame: a nitrous oxide-acetylene flame and a 1000 ppm Al solution as ionization buffer were used for Mo analyses. Background corrections were made on Ni, Cd, Ag, Pb and Mo analyses.

F was determined using an ion selective electrode and total ionic strength adjustment buffer (TISAB). A 500 mg sample was fused with Na<sub>2</sub>O<sub>3</sub>-KNO<sub>3</sub> (9:1) at 800° C for 10 minutes.

Ba was analysed by x-ray fluorescence using pressed powder disks.

Colorimetric methods employing the standard dithiol method were used to measure W. The sample was fused at 800° C in nickel crucibles along with a mixture of  $Na_2CO_3$ , NaCl and KNO<sub>3</sub> (5:4:1).

 $P_2O_5$  was determined colorimetrically using the  $H_2SO_4$  and ammonium molybdate method after sample decomposition using  $HNO_3$ - $HCI-HCIO_4$ .

Loss-on-ignition (LOI) was recorded as a measure of the weight percent loss of volatiles after igniting a known weight at 450° C for four hours.

U was determined by neutron activation analysis using delayed neutron counting on a 3 g sample. The flux density used was  $2 \times 10^{11}$  neutrons/cm<sup>2</sup>/s to  $1 \times 10^{12}$  neutrons/cm<sup>2</sup>/s.

For the determination of As, Sb and Hg, a 750 ml sample was digested overnight in aqua regia (3 HCI:1 HNO3). It was then treated in a water bath for approximately four hours during which time the temperature was raised to 90° C. The final volume was 15 ml. Subsequently, As and Sb were measured by atomic absorption spectrophotometry using a silica tube heated to 900° C. The hydride of the element was formed with a sodium borohydride reducing agent. Mercury was determined similarly by measuring Hg vapour atomized in a silica tube heated to 100° C.

## Instrumental Neutron Activation Analysis (INAA)

Weighed and encapsulated samples are packaged for irradiation along with internal standards and international reference materials. Samples and standards are irradiated together with neutron flux monitors in a two-megawatt pool-type reactor. After a seven-day decay period, samples are measured on a high-resolution germanium detector. Computer control is achieved with a Microvax II computer. Typical counting times are 500 seconds. Elements determined by INAA include: Ag, As, Au, Ba, Br, Cd, Ce, Co, Cr, Cs, Eu, Fe, Hf, Ir, La, Lu, Mo, Na, Ni, Rb, Sb, Sc, Se, Sm, Sn, Ta, Tb, Te, Th, U, W, Yb, Zn, and Zr. The sample weights are also reported. Data for Ag, Cd, Ir, Mo, Ni, Se, Sn, Te, Zn, and Zr are not published because of inadequate detection limits and/or precision.

Table 2 provides a summary of lower detection limits and methods for elements determined in stream sediments.

ELEMENT		DETECTION LEVEL		METHOD	
SEDIMENTS					
	Silver		02	nnm	AAS
Δs	Arsenic		0.4	ppm	HY-AAS
As	Arsenic		0.5	ppm	INAA
Au	Gold		2	pphi	INAA
Ba	Barium		0.02	pct	XRF
Ва	Barium		50	maa	INAA
Br	Bromine		0.5	mag	INAA
Cd	Cadmium		0.2	maa	AAS
Ce	Cerium		5	maa	INAA
Co	Cobalt		2	ppm	AAS
Co	Cobalt		5	ppm	INAA
Cr	Chromium		20	ppm	INAA
Cs	Caesium		0.5	ppm	INAA
Cu	Copper		2	mqq	AAS
Eu	Europium		1	ppm	INAA
F	Fluorine		20	ppm	ISE
Fe	Iron		0.2	pct	AAS
Fe	Iron		0.2	pct	INAA
Hf	Hafnium		1	, ppm	INAA
Hg	Mercury		30	ppb	CV-AAS
La	Lanthanum		2	ppm	INAA
LOI	Loss-on-ignition		1.0	pct	GRAV
Lu	Lutetium		0.02	ppm	INAA
Mn	Manganese		2	ppm	AAS
Мо	Molybdenum		2	ppm	AAS
Na	Sodium		0.02	pct	INAA
Ni	Nickel		2	ppm	AAS
$P_2O_5$	Phosphoric anydride		0.04	pct	COL
Pb	Lead		2	ppm	AAS
Rb	Rubidium		5	ppm	INAA
Sb	Antimony		0.4	ppm	HY-AAS
Sb	Antimony		0.1	ppm	INAA
Sc	Scandium		0.2	ppm	INAA
Sm	Samarium		0.1	ppm	INAA
Та	Tantalum		0.5	ppm	INAA
Tb	Terbium		0.5	ppm	INAA
Th	Thorium		0.2	ppm	INAA
U	Uranium		1.0	ppm	NADNC
U	Uranium		0.2	ppm	INAA
V	Vanadium		20	ppm	AAS
W	Tungsten		2	ppm	COL
W	lungsten		1	ppm	INAA
Yb	Ytterbium		2	ppm	INAA
Zn	Zinc		2	ppm	AAS
AuWt	Sample weight		0.01	g	-

AAS - atomic absorption spectrometry

COL - colorimetric methods CV-AAS - cold vapour / atomic absorption spectrometry

FUS - fusion GRAV - gravimetry HY-AAS - hydride evolution followed by atomic absorption spectrometry INAA - Instrumental Neutron Activation Analysis ISE - ion selective electrode

NADNC - neutron activation followed by delayed neutron counting

XRF - x-ray fluorescence

Table 2

Summary of analytical data and methods for sediments

# Acknowledgments

W.D. Goodfellow, GSC, directed a regional stream sediment and water survey carried out in 1981 by Marshall, Macklin and Monaghan Ltd. Sediment samples were prepared for analysis by staff at the Geological Survey of Canada. Bondar-Clegg and Company, Ottawa, determined the following elements in stream sediments: Zn, Cu, Pb, Ni, Co, Ag, Mn, Fe, Mo, V, W, F, and Ba. Loss-on-ignition (LOI) and P2O5 were also determined. Nova Track Ltd., Vancouver, measured U in stream sediments using a delayed neutron activation method. A. I. MacLaurin, G. Gauthier and W. M. Alexander carried out the rest of the analytical work under the supervision of G. E. M. Hall. J.J. Lynch monitored analytical results.

Stream sediment samples were reanalysed by Becquerel Labs, Mississauga, Ontario, under the direction of P.W. Friske. The authors would like to thank W. Spirito for critically reviewing the open file and offering many helpful suggestions.

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